

A Comprehensive Study of Thermo-Physical Properties of Investment Shells

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Outline

- Thermal properties measurements
- Process history effect on silica based shells

Introduction

Thermal Properties of shells

- Casting microstructures
- Casting defects
- > Solidification simulations

Challenges

- > Various levels of porosities
- Metastable components

Before this work...

- Properties from handbook
- Rule of mixture estimation considering shell porosity

$$k_r = \exp(-1.5\varphi/(1-\varphi))$$

OHot wire method

Standard Laser Flash

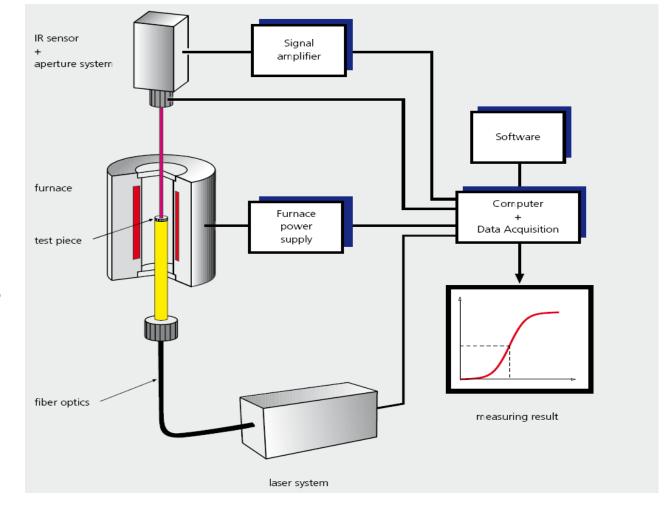
✓ Temperature dependent Diffusivity (α):

$$\alpha = 0.1388 L^2/t_{1/2}$$

L: specimen thickness $t_{1/2}$: time required for rear face temperature to reach 50% of its maximal value

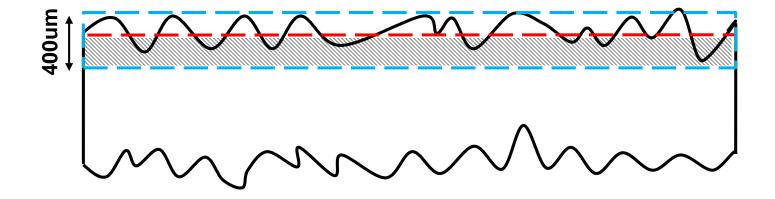
$$\checkmark \quad \mathbf{C}_{p}: \qquad \left(\rho c_{p}\right)_{M} = \frac{L_{R} \Delta T_{R}}{L_{M} \Delta T_{M}} \left(\rho c_{p}\right)_{R}$$

✓ Thermal conductivity (*K*):

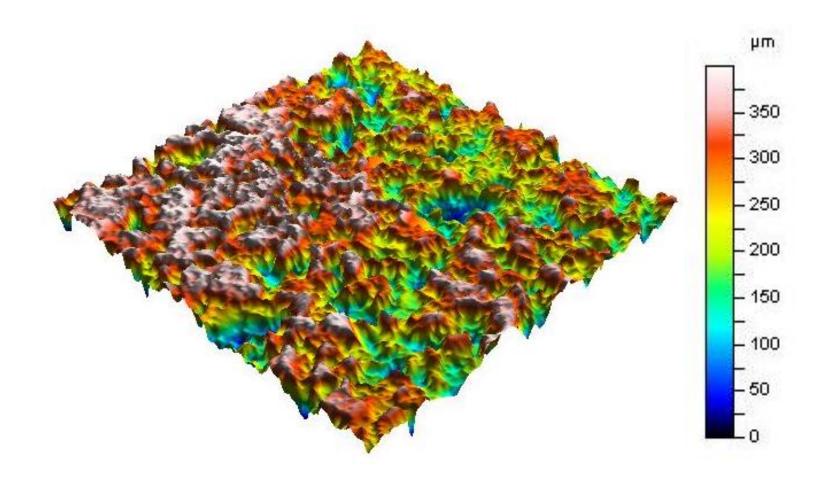


Surface roughness

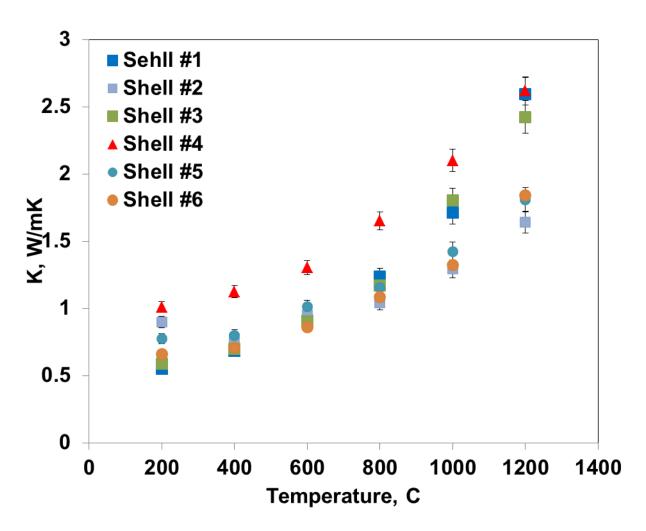
- Challenges to accurately determine the sample thickness
- "effective" plane is drawn
 - Volume of "mountain" = volume of "valley"



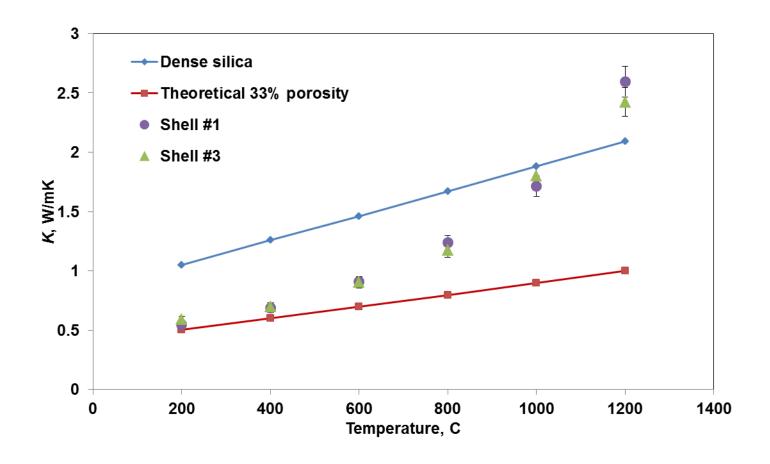
Effective Thickness Using 3D Optical Profiler



Thermal Conductivity from Laser Flash



Comparison to Theoretical values



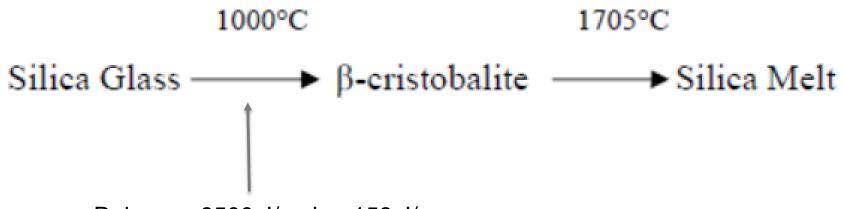
Discrepancy from photon conductivity

Time to think...

- Modified laser flash
 - Samples equilibrated at test temperatures
 - Good for steady state thermal properties

In Reality...

Phase transformation of fused silica



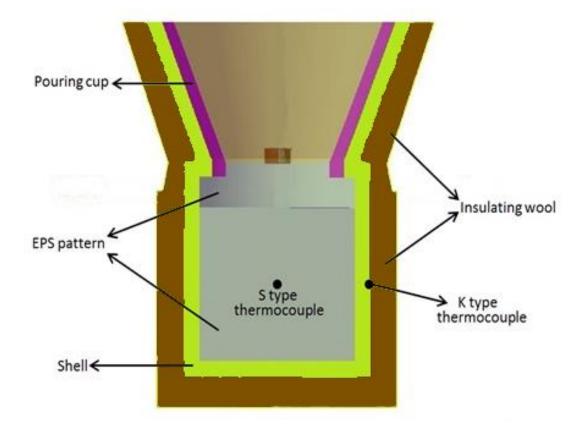
Releases 9500 J/mol or 158 J/g

• Ways to measure the transient properties?

Inverse Method

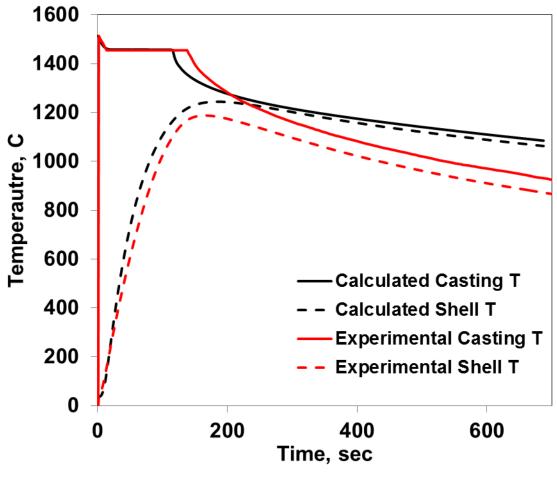
Calculating thermal properties from a real-time heat

- The shell was pre-fired at 850 C for 1 hour and cooled to room temperature
- Reference material pure Ni was poured at 1520 C into the shell
- Two temperature curves were recorded

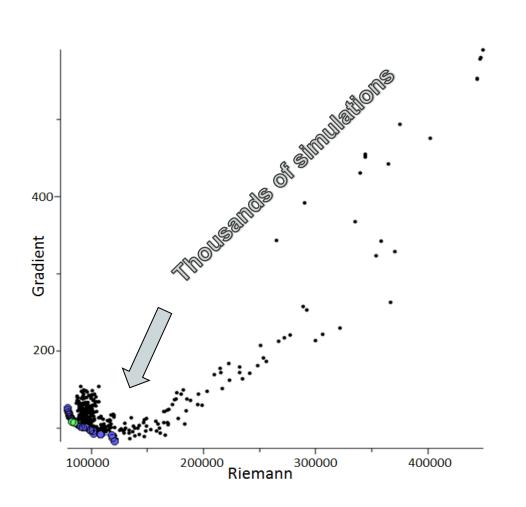


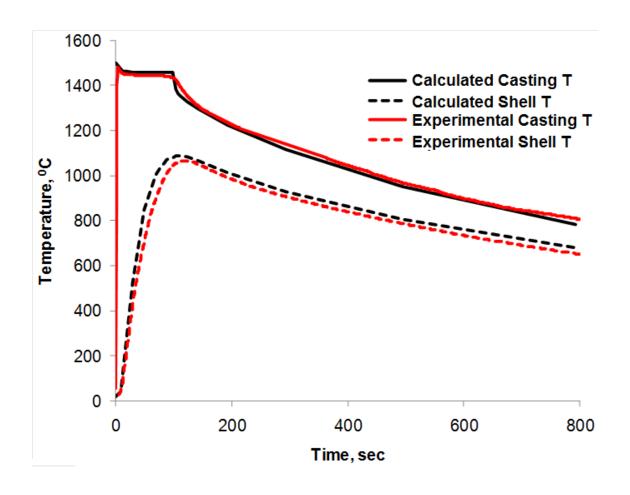
Inverse Method

OLF results as the starting point

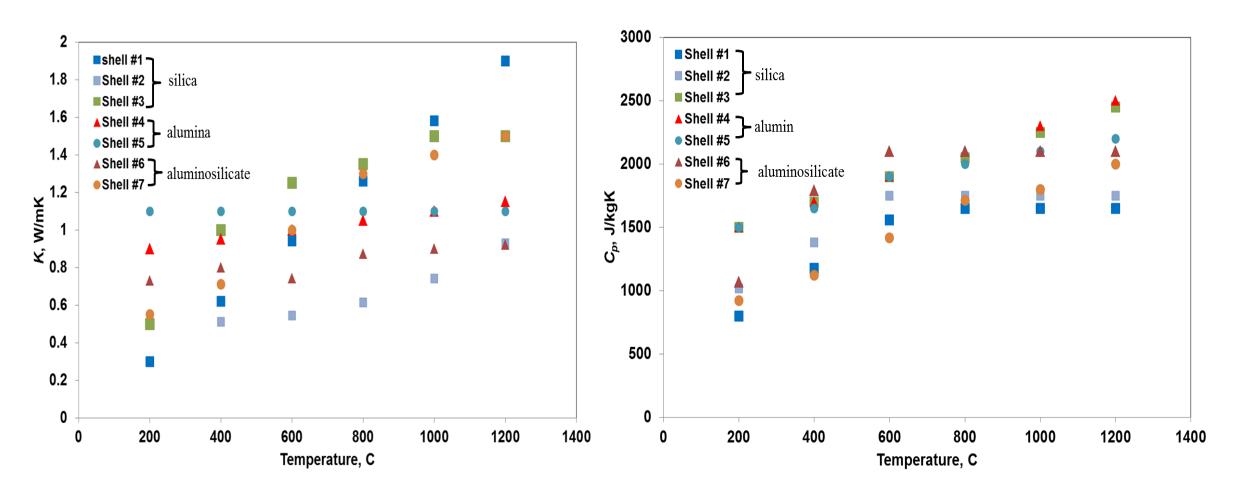


Inverse Optimization



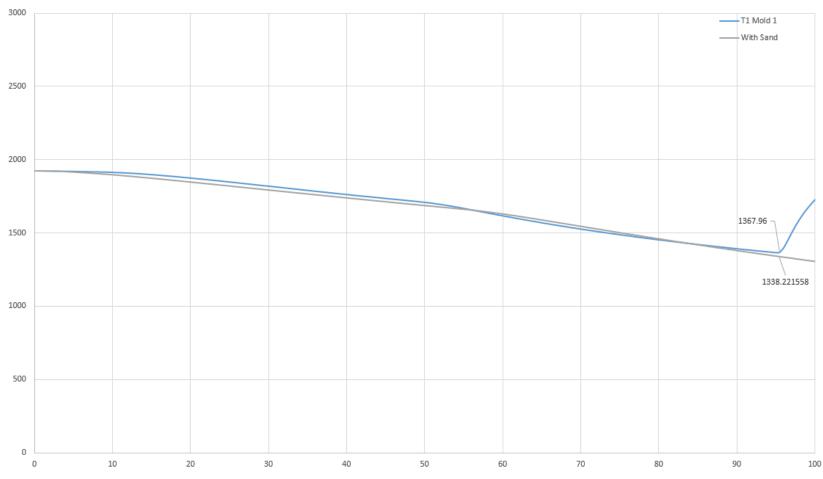


Examples for database



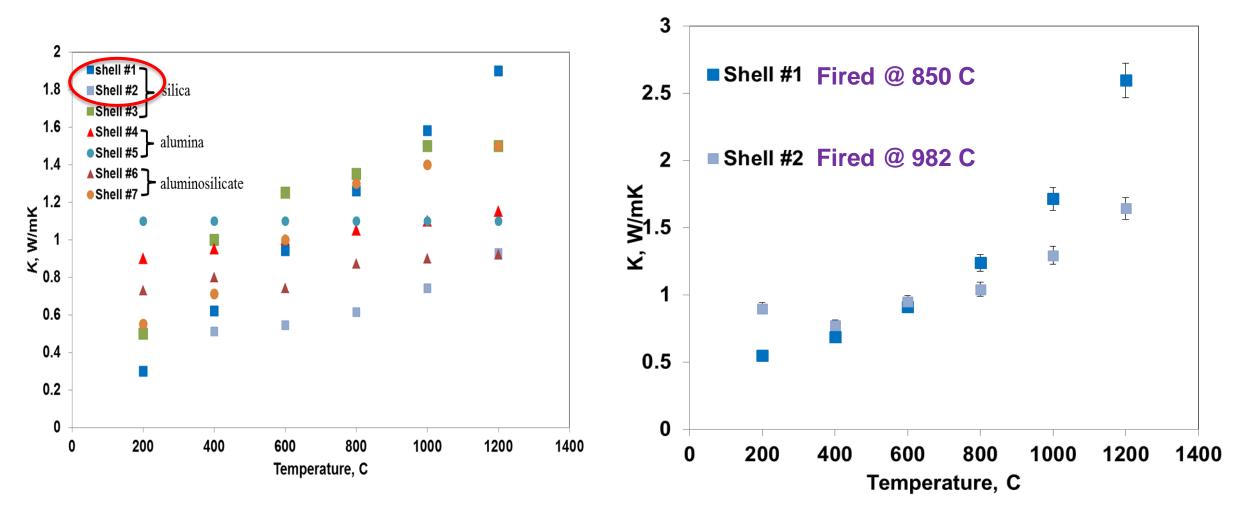
Xu et al., International Journal of Metalcasting, Volume 10, Issue 3, 2016

Industrial Responses



"Overall I think we are very satisfied with the results of this test, I think the material properties you calculated were spot on."

Time to think again...



Question: different firing parameters result in different amount of devitrification?

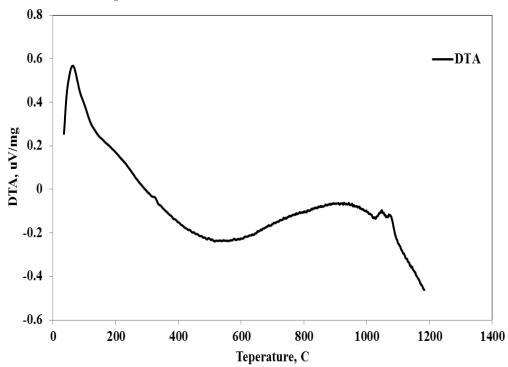
Shell Preparation

- Fused silica based shell
- Fired at 600°C, 850°C, 1000°C for one hour
- 3-point bend samples
- Shell ingredients dried and crushed

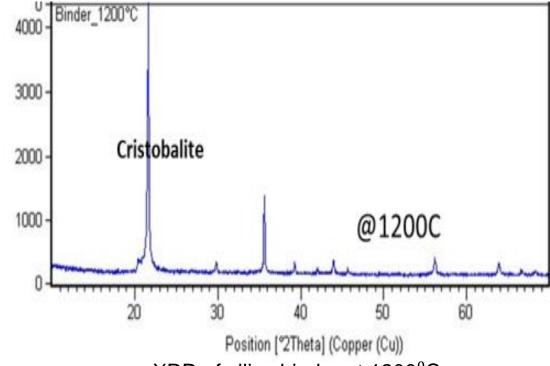


Silica Devitrification

 Silica binder was found the only ingredient that devitrifies at up to 1200°C



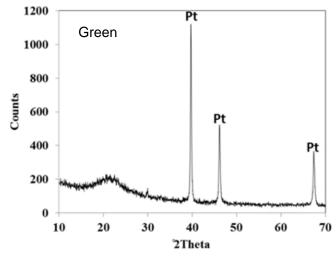
DTA at a 15°C/min heating rate

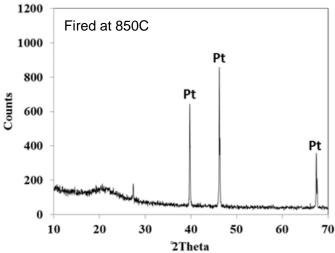


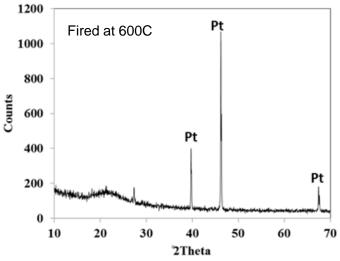
XRD of silica binder at 1200°C

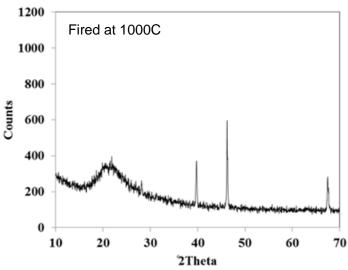
Phases Present after Firing

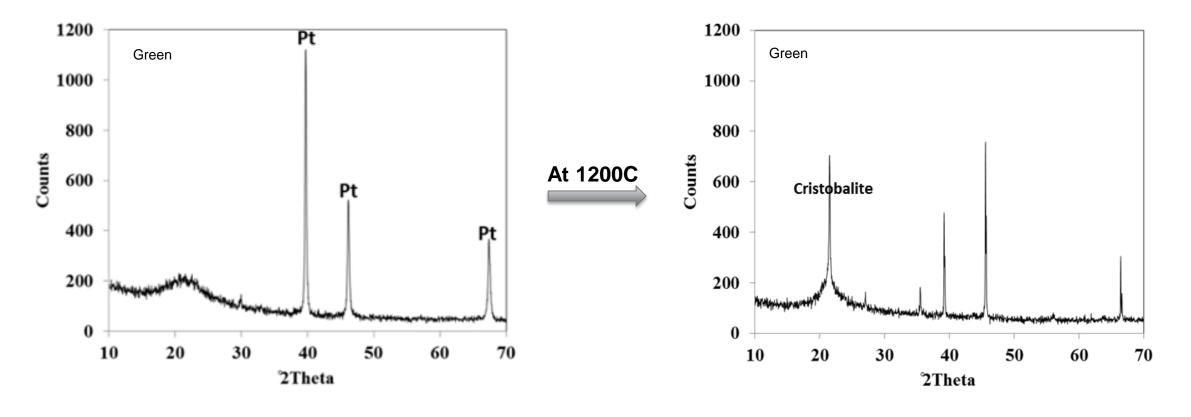
● Fired at 600°C, 850°C, 1000°C for one hour – XRD at RT

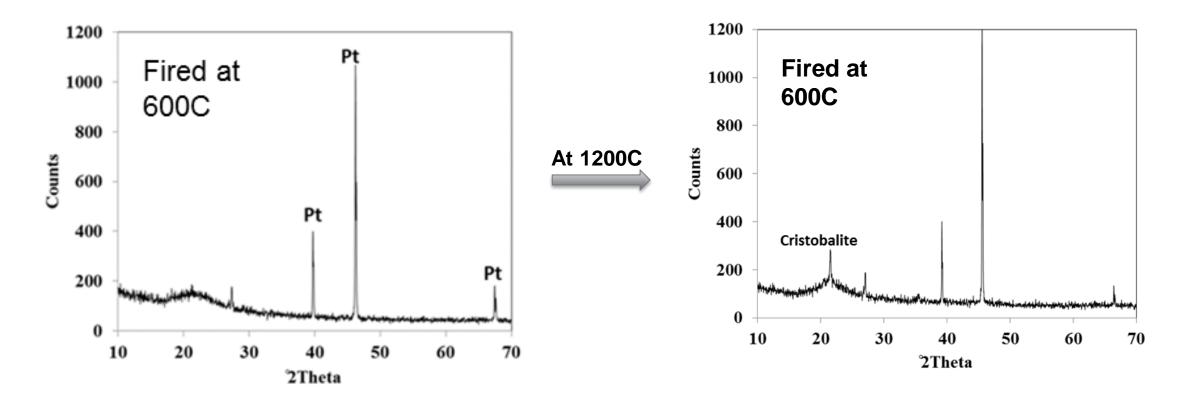


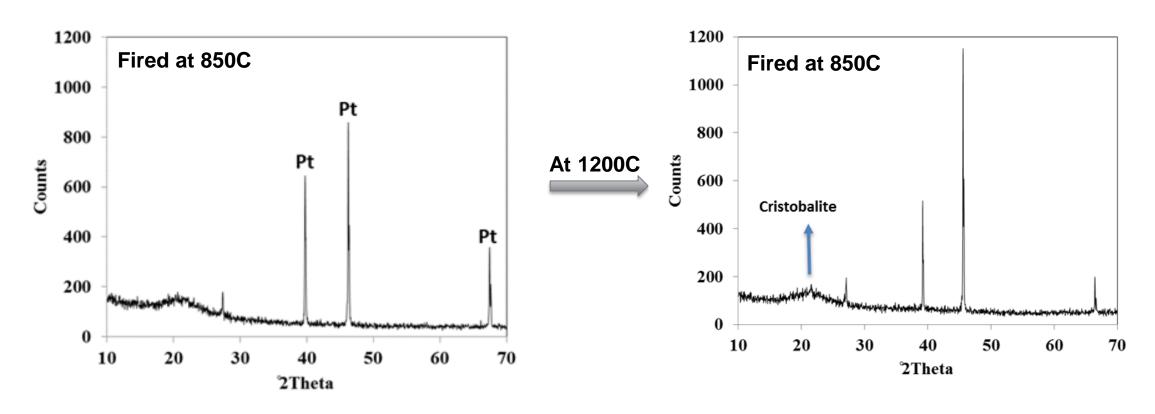


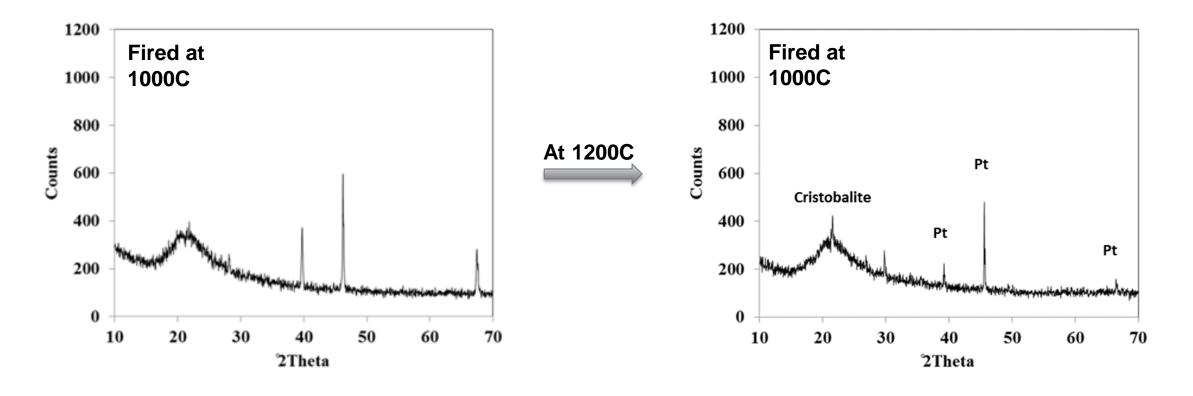












What Caused the Change

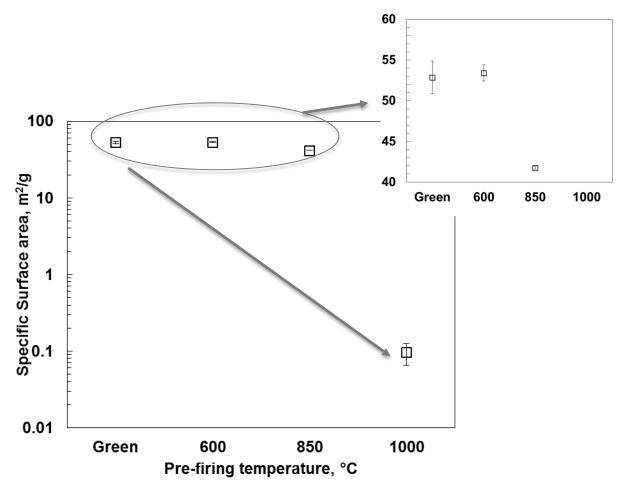
During firing/sintering

- Grain boundary migration
- Grain growth



- Surface area reduction
 - > Surface energy to activate
 - Kinetical favorable path

Specific Surface Area of Silica Binder

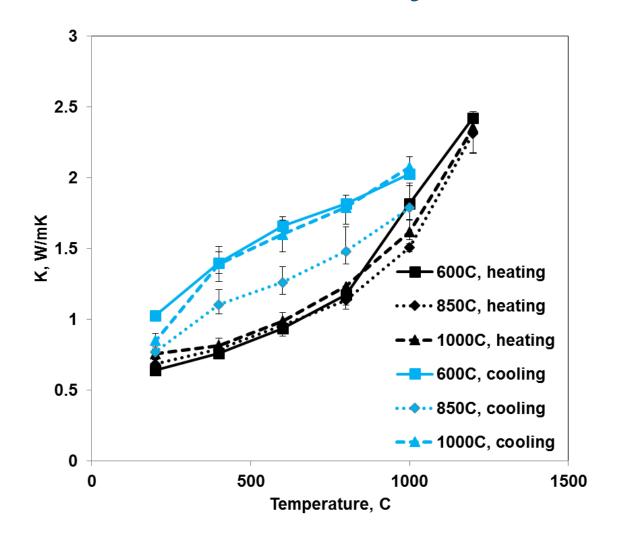


0.26J/m² surface energy* ~1,000 J/g surface energy loss

*Brunauer, S., Kantro, D. L., Weise, C. H., "The Surface Energies of Amorphous Silica and Hydrous Amorphous Silica", Canadian Journal of Chemistry, 1956, 34(10): 1483-1496, 10.1139/v56-190.

Thermal Conductivity

- Crystalline silica has better thermal conductivity
- Amount of crystalline silica
 - > 600C > 1000 > 850C



Summary

- Ways to determine thermal properties
 - Modified LF: steady-state thermal properties
 - > Inverse method: transit
- Firing parameters have a significant effect on thermal properties

Current Work

- Funded by American Foundry Society
 - > Develop silica devitrification kinetical model
- Future work
 - >Silica reclamation

Acknowledgements

- My wife Dr. Jingjing Qing
- Dr. Von Richards, Dr. Simon Lekakh
- Shelly Dutler
- Participating industrial foundries
- Undergraduate assistances at S&T

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